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(21) International Application Number: PCT/US96/17943 (22) International Filing Date: 8 November 1996 (08.11.96) (30) Priority Data: 08/552,506 9 November 1995 (09.11.95) US (71) Applicant: BIOMETRIC IMAGING, INC. [US/US]; 1025 Bella Avenue, Mountain View, CA 94043 (US). (72) Inventor: LEE, Linda, G.; 3187 Stelling Drive, Palo Alto, CA 94303 (US). (74) Agents: PARKHURST, David, G. et al.; Fulwider Patton Lee & Utecht, L.L.P., 10th floor, 10877 Wilshire Boulevard, Los Angeles, CA 90024 (US).		(81) Designated States: AL, AM, AT, AU, AZ, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IL, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, UZ, VN, ARIPO patent (KE, LS, MW, SD, SZ, UG), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG). Published <i>With international search report.</i>	
(54) Title: CYCLIZED FLUORESCENT DNA-INTERCALATING CYANINE DYES (57) Abstract <p>New intercalating asymmetric cyanine dyes are provided in which the benzothiazole portion of the cyanine dye has been modified to produce dyes with improved properties for labelling nucleic acids, such as longer wavelengths and improved fluorescence enhancement when bound to DNA or RNA. More specifically, the dyes are cyclized fluorescent cyanine dyes for non-covalently labelling nucleic acids. Methods are described for detecting nucleic acids in a sample by contacting the nucleic acids with a fluorescent cyanine dye and monitoring the change in fluorescence emission of the dye.</p>			

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Title: CYCLIZED FLUORESCENT DNA-INTERCALATING CYANINE DYES**FIELD OF THE INVENTION**

The present invention relates generally to dyes for labelling nucleic acids. More specifically, the present invention relates to intercalating cyanine dyes for the detection and enumeration of nucleic acids.

5 BACKGROUND OF THE INVENTION

Intercalating dyes which exhibit enhanced fluorescence upon binding to DNA or RNA are a basic tool in molecular and cell biology. In general, intercalating dyes bind noncovalently to DNA through a combination of hydrophobic interactions with the DNA base-pairs and ionic binding to the negatively charged phosphate backbone. The fluorescence of the dye is ideally increased several-fold upon binding to DNA, thereby enabling the detection of small amounts of nucleic acids. Examples of fluorescent noncovalent DNA binding dyes include ethidium bromide which is commonly used to stain DNA in agarose gels after gel electrophoresis, and propidium iodide and Hoechst 33258 which are used in flow cytometry to determine the DNA ploidy of cells.

Fluorescent nucleic acid labelling dyes preferably absorb light between about 300 and 900 nm and preferably have a Stokes shift of at least about 10 nm. Dyes that absorb light in the 500 to 900 nm range are preferred because they are spectrally removed from other components that may be present in a biological sample and because they may be used with inexpensive light sources. Fluorescent dyes that have a high extinction coefficient, a high quantum yield, and significantly enhanced fluorescence when bound to a nucleic acid are also preferred.

25 Few new dye chromophores were described until the introduction of Thiazole Orange as a reticulocyte stain in 1986. Lee, et al., "Thiazole Orange: A New Dye for Reticulocyte Analysis", Cytometry 1986 7, 508-517. Thiazole

Orange is an asymmetric cyanine dye. Although many asymmetric cyanine dyes have been described in the art (e.g., Lincoln, et al., U.S. Patent No. 3,282,932), Thiazole Orange's fluorescence properties when bound to DNA and RNA and its utility for labelling nucleic acids had not been previously identified. Lee, et al., U.S. Patent No. 4,957,870. For example, unlike most asymmetric cyanine dyes, Thiazole Orange exhibits a several thousand-fold enhancement in fluorescence upon binding to DNA.

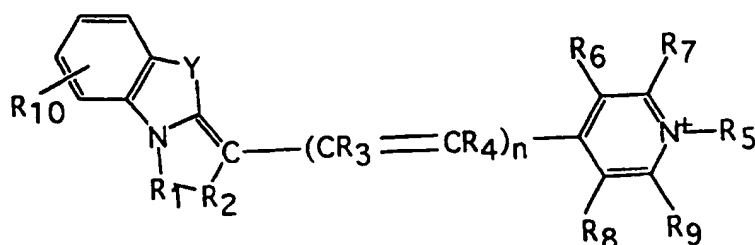
Since the discovery of Thiazole Orange as a nucleic acid dye, several improvements to Thiazole Orange and its trimethine homologs have been developed to provide dyes with tighter binding to DNA and greater water solubility. Xue, et al. U.S. Patent No. 5,321,130 and Glazer, et al. U.S. Patent No. 5,312,921. These dyes generally involve a modification to the quinolinium portion of the dye.

A continuing need exists for new and improved dyes for labelling nucleic acids. In particular, a need exists for dyes which exhibit longer wavelengths and significantly enhanced fluorescence when bound to DNA or RNA.

SUMMARY OF THE INVENTION

The present invention relates asymmetric cyanine dyes for non-covalently labelling nucleic acids in which the benzothiazole portion of the dye has been modified to provide improved physical properties to the dye, such as longer wavelengths and improved fluorescence enhancement when bound to DNA or RNA.

More specifically, the invention relates to cyclized fluorescent cyanine dyes for non-covalently labelling nucleic acids. The cyclized fluorescent cyanine dyes according to the present invention are represented by General Formula I



where:

n is 0, 1 or 2;

Y may be either S or O;

R₁ and R₂ are taken together to form a 5, 6, 7 or 8 membered ring;

R₃ and R₄ may each independently be either hydrogen, C₁ - C₁₀ alkyl, C₁ - C₁₀ alkoxy, or C₁ - C₁₀ alkylthio;

R₅ may be a C₁ - C₃₀ alkyl, preferably substituted with one or more polar substituents which preferably includes one or more positively charged atoms, or a cyclized fluorescent cyanine dye of the

present invention, i.e., where R_5 is a linker between two cyclized fluorescent cyanine dyes;

R_6 and R_7 may each independently be either H or C_{1-10} alkyl, or may be taken together to form a 5 or 6 membered ring, most preferably a 6 membered aromatic ring, optionally substituted with C_{1-6} alkyl or $C_1 - C_{10}$ alkoxy groups;

R_8 and R_9 may each independently be either H or C_{1-10} alkyl, or may be taken together to form a 5 or 6 membered ring, most preferably a 6 membered aromatic ring, optionally substituted with C_{1-6} alkyl or $C_1 - C_{10}$ alkoxy groups; and

R_{10} may be either H, C_{1-6} alkyl, $C_1 - C_{10}$ alkoxy or a fused benzene.

As used above, alkyl and alkoxy refer to any substituent having a carbon backbone having the specified range of carbon atoms. The carbon backbone may form a straight chain, may be branched or may be cyclic. The alkyl and alkoxy groups may be optionally substituted by a wide variety of substituents including, for example, alcohols, amines, thiols, phosphates, halides, ethers, esters, ketones, aldehydes, carboxylic acids, amides, cycloalkyls, and aromatic rings.

The invention also relates to the composition of a cyanine dye according to the present invention non-covalently bound to a nucleic acid sequence, i.e., RNA or DNA, which enables the nucleic acid sequence to be analytically detected.

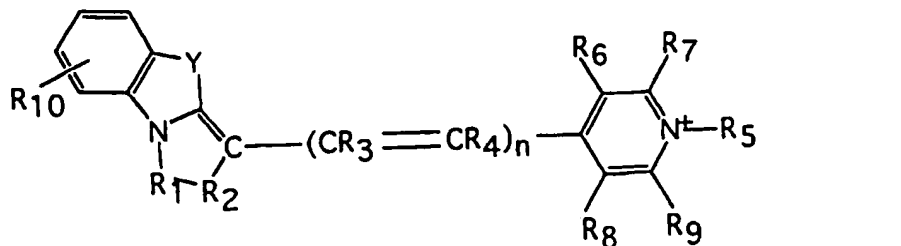
The invention also relates to a method for detecting nucleic acids in a sample by contacting the nucleic acids with a fluorescent cyanine dye according to the present invention and monitoring the change in fluorescence emission of the dye.

DETAILED DESCRIPTION OF THE INVENTION

The present invention relates asymmetric cyanine dyes for non-covalently labelling nucleic acids in which the benzothiazole portion of the dye

has been modified to provide improved physical properties to the dye, such as longer wavelengths and improved fluorescence enhancement when bound to DNA or RNA.

In one embodiment, the present invention relates to cyclized fluorescent cyanine dyes generally represented by General Formula I



where:

n is 0, 1 or 2;

Y may be either S or O;

R_1 and R_2 are taken together to form a 5, 6, 7 or 8 membered ring;

R_3 and R_4 may each independently be either hydrogen, $C_1 - C_{10}$ alkyl, $C_1 - C_{10}$ alkoxy, or $C_1 - C_{10}$ alkylthio;

R_5 may be a $C_1 - C_{50}$ alkyl, preferably substituted with one or more polar substituents which preferably includes one or more positively charged atoms, or a cyclized fluorescent cyanine dye of the present invention, i.e., where R_5 is a linker between two cyclized fluorescent cyanine dyes;

R_6 and R_7 may each independently be either H or C_{1-10} alkyl, or may be taken together to form a 5 or 6 membered ring, most preferably a 6 membered aromatic ring, optionally substituted with C_{1-6} alkyl or $C_1 - C_{10}$ alkoxy groups;

R_8 and R_9 may each independently be either H or C_{1-10} alkyl, or may be taken together to form a 5 or 6 membered ring, most preferably a 6

membered aromatic ring, optionally substituted with C₁₋₆ alkyl or C₁ - C₁₀ alkoxy groups; and

R₁₀ may be either H, C₁₋₆ alkyl, C₁ - C₁₀ alkoxy or a fused benzene.

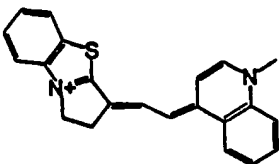
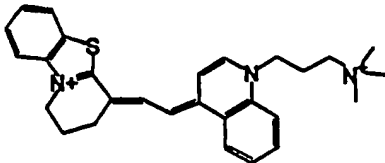
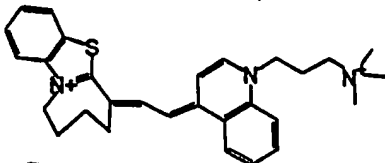
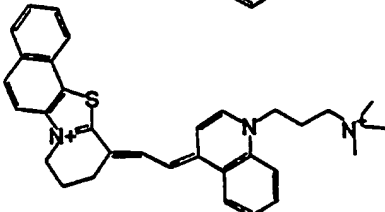
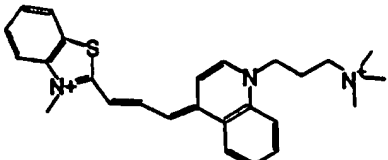
5 As used above, alkyl and alkoxy refer to any substituent having a carbon backbone having the specified range of carbon atoms. The carbon backbone may form a straight chain, may be branched or may be cyclic. The alkyl and alkoxy groups may be optionally substituted by a wide variety of substituents including, for example, alcohols, amines, thiols, phosphates, halides, ethers,
10 esters, ketones, aldehydes, carboxylic acids, amides, cycloalkyls, and aromatic rings.

 The cyclized cyanine dyes of the present invention provide the advantage over previous cyanine dyes of having higher absorbance and emission wavelengths. The cyclized cyanine dyes preferably absorb light at a
15 wavelength of at least about 640 nm, more preferably at least about 649 nm and emit fluorescence at a wavelength of at least about 650 nm, more preferably at least about 663 nm. The cyclized cyanine dyes also preferably have a positive Stoke's shift ($\lambda_{\text{Emission}} - \lambda_{\text{Abs.}}$) of at least about 12 nm.

 In particular, cyclized cyanine dyes having General Formula I where R₁
20 and R₂ are taken together to form a 5, 6, 7 or 8 membered ring have been found to absorb light and fluoresce when bound to a nucleic acid polymer at unexpectedly higher wavelengths than has been previously achieved by cyanine dyes where R₁ and R₂ do not form a ring structure.

 Fluorescent cyanine dyes having the General Formula I where R₁ and R₂
25 are taken together to form a 7 membered ring have also been found to have the greatest Stoke's shift ($\lambda_{\text{Emission}} - \lambda_{\text{Abs.}}$)

TABLE 1: Absorbance and Emission Maxima of Intercalating Dyes in PBS with Excess DNA ([bp]/[dye] = 100)

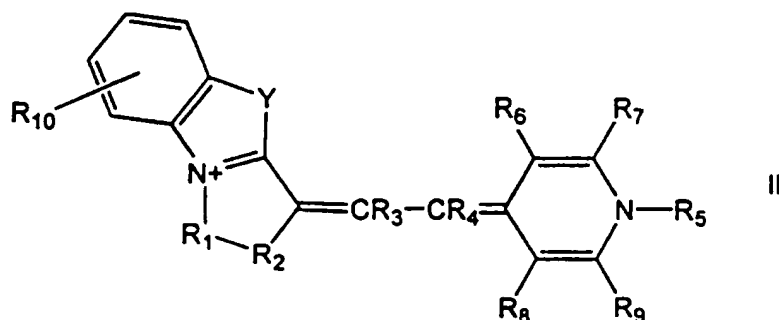
<u>COMPOUND</u>		<u>Abs_{max}</u>	<u>Ems_{max}</u>	<u>F.E.</u>	
5		1	649	663	100X
		2	654	667	100X
		3	654	672	30X
		4	675	690	200X
		5*	641	655	100X

- 10 Abs_{max}-Absorbance maximum (bounds to DNA)
Ems_{max}-Emission maximum (bound to DNA)
F.E.-fluorescence enhancement (bounds vs. not bound to DNA or RNA)
* Compound 5 is taught in U.S. Patent No. 5,321,130 to Yue, et al.

Table 1 summarizes the absorbance maximum and fluorescence emission maximum wavelengths (both when bound to DNA) of some exemplary cyclized cyanine dyes of the present invention.

As illustrated in Table 1, it was found that the addition of a cyclic aliphatic side chain to the basic cyanine dye structure, i.e., formation of a 5-8 membered ring by combining R_1 and R_2 , was found to increase the absorbance and fluorescence emission wavelengths of the corresponding acyclic cyanine dye by about 12 nm. For example, as shown with regard to dyes 2 and 5, dye 2 has an Abs_{max} at 654 nm as compared to 641 nm and an Ems_{max} at 667 nm as compared to 655 nm. In addition, dye 4 is the longest wavelength trimethine intercalating dye yet reported.

With regard to n , n may equal 1. Accordingly, the present invention includes cyclized cyanine dyes having the General Formula II (i.e. where $n = 1$)



where Y , R_1 , R_2 , R_3 , R_4 , R_5 , R_6 , R_7 , R_8 , R_9 and R_{10} are as specified above.

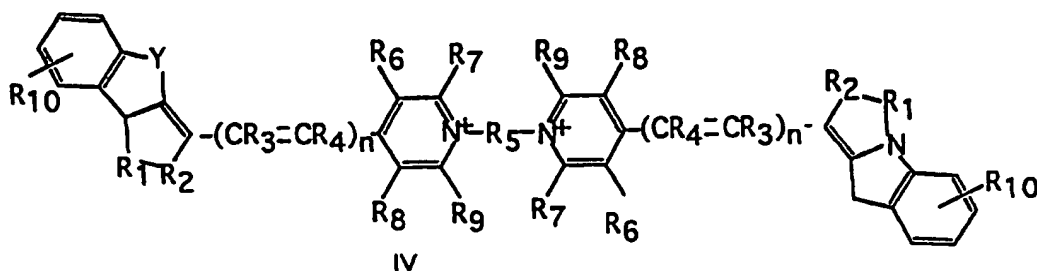
Y may be either S or O, and is most preferably S.

R_3 and R_4 may each independently be either hydrogen, $C_1 - C_{10}$ alkyl, $C_1 - C_{10}$ alkoxy, or $C_1 - C_{10}$ alkylthio, and are preferably H.

R_5 may be a $C_1 - C_{50}$ alkyl. Since DNA and RNA to which the cyclized cyanine dyes bind contain negatively charged phosphate backbones, it is preferred that R_5 be substituted with one or more polar substituents. It is most preferred that R_5 include one or more positively charged atoms in the polar

substituent. U.S. Patent No. 5,321,130 to Yue, et al. teaches unsymmetrical cyanine dyes having an aminoalkyl chain containing a backbone of 3-42 carbons and 1-5 positively charged nitrogen atoms. The cationic tail described in U.S. Patent No. 5,321,130 exemplifies one of the positively charged R_5 substituents that may be used in combination with the cyclic cyanine dyes of the present invention and is incorporated herein by reference. In addition to the positively charged R_5 substituents described in U.S. Patent No. 5,321,130, R_{12} is also intended to include aminoalkyl chains including a positively charged cyclic aminoalkyl group having 1-5 positively charged nitrogen atoms.

Alternatively, R_5 may form part of a linker between two cyclized fluorescent cyanine dyes as illustrated by General Formula IV



According to this embodiment, Y, R_1 , R_2 , R_3 , R_4 , R_6 , R_7 , R_8 , R_9 and R_{10} are as specified above. It should be noted that the two linked cyanine dyes may be the same or different cyanine dyes. In general, it is preferred that the linked cyanine dyes be the same since different dyes will have different spectral properties.

R_6 and R_7 may each independently be either H, C_{1-10} alkyl, or are taken together to form a 5 or 6 membered ring, most preferably a 5 or 6 membered aromatic ring, optionally substituted with C_{1-4} alkyl or $C_1 - C_{10}$ alkoxy groups.

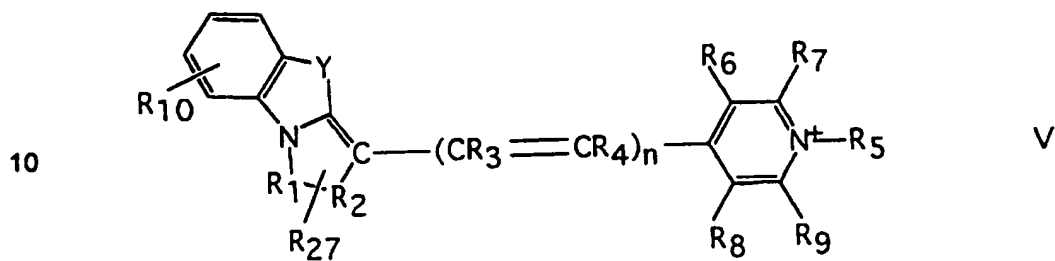
R_8 and R_9 may each independently be either H, C_{1-10} alkyl, or are taken together to form a 5 or 6 membered ring, most preferably a 5 or 6 membered aromatic ring, optionally substituted with C_{1-4} alkyl or $C_1 - C_{10}$ alkoxy groups.

In general, it is preferred either R_6 and R_7 or R_8 and R_9 are taken together to form a 5 or 6 membered aromatic ring, optionally substituted with

C₁₋₆ alkyl or C₁ - C₁₀ alkoxy groups. The R₆ and R₇ or R₈ and R₉ groups that do not form the aromatic ring are preferably H.

R₁₀ may be either H, C₁₋₆ alkyl, C₁ - C₁₀ alkoxy or a fused benzene.

5 In a particularly preferred embodiment, the cyclized cyanine dye has the General Formula V where the ring formed by R₁ and R₂ includes a positively charged substituent R₂₇. As discussed herein, inclusion of a positively charged substituent, such as R₂₇, to a substituent on the positively charged nitrogen on the benzothiazole ring improves the net fluorescence enhancement of the dye with DNA.



R₂₇ is a positively charged alkyl substituent which may be attached to any atom used to form the 5, 6, 7 or 8 membered ring. R₂₇ is more preferably a positively charged aminoalkyl substituent. For example, R₁₂ can be an aminoalkyl chain containing a backbone of 3-42 carbons and 1-5 positively charged nitrogen atoms as described in U.S. Patent No. 5,321,130 to Yue, et al. which is incorporated herein by reference. In addition to the positively charged substituents described in U.S. Patent No. 5,321,130, R₁₂ is also intended to include aminoalkyl chains including a positively charged cyclic aminoalkyl group having 1-5 positively charged nitrogen atoms.

15

In a preferred embodiment, R_{27} has the general formula $-R_{28}N(R_{29}R_{30}R_{31})$ where R_{28} is a C_{1-5} alkyl and R_{29} , R_{30} , and R_{31} are each independently a C_{1-10} alkyl.

5 Table 2 provides examples of some of the preferred cyclized cyanine dyes. It should be understood, however, that the dyes listed in Table 2 are intended only to exemplify the cyclized cyanine dyes of the present invention and are not intended to be limiting.

TABLE 2

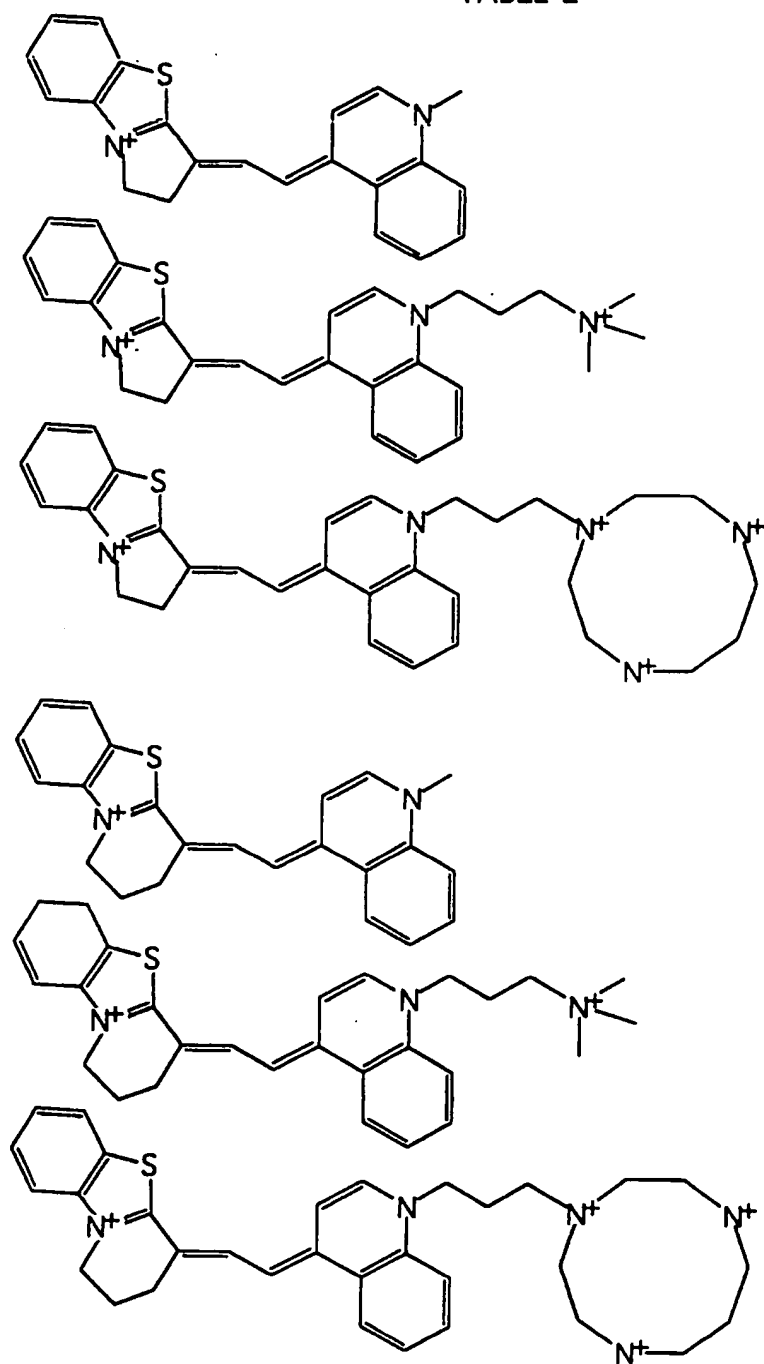


TABLE 2 (cont.)

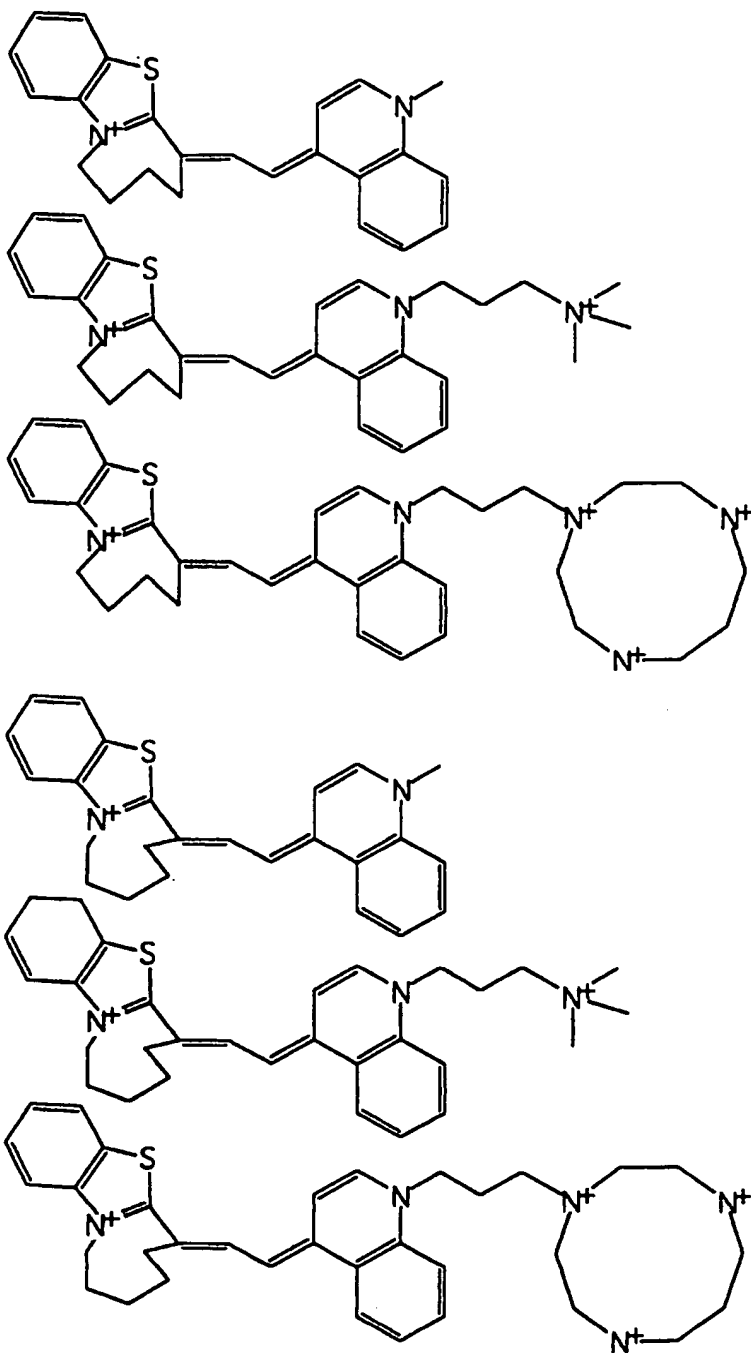


TABLE 2 (cont.)

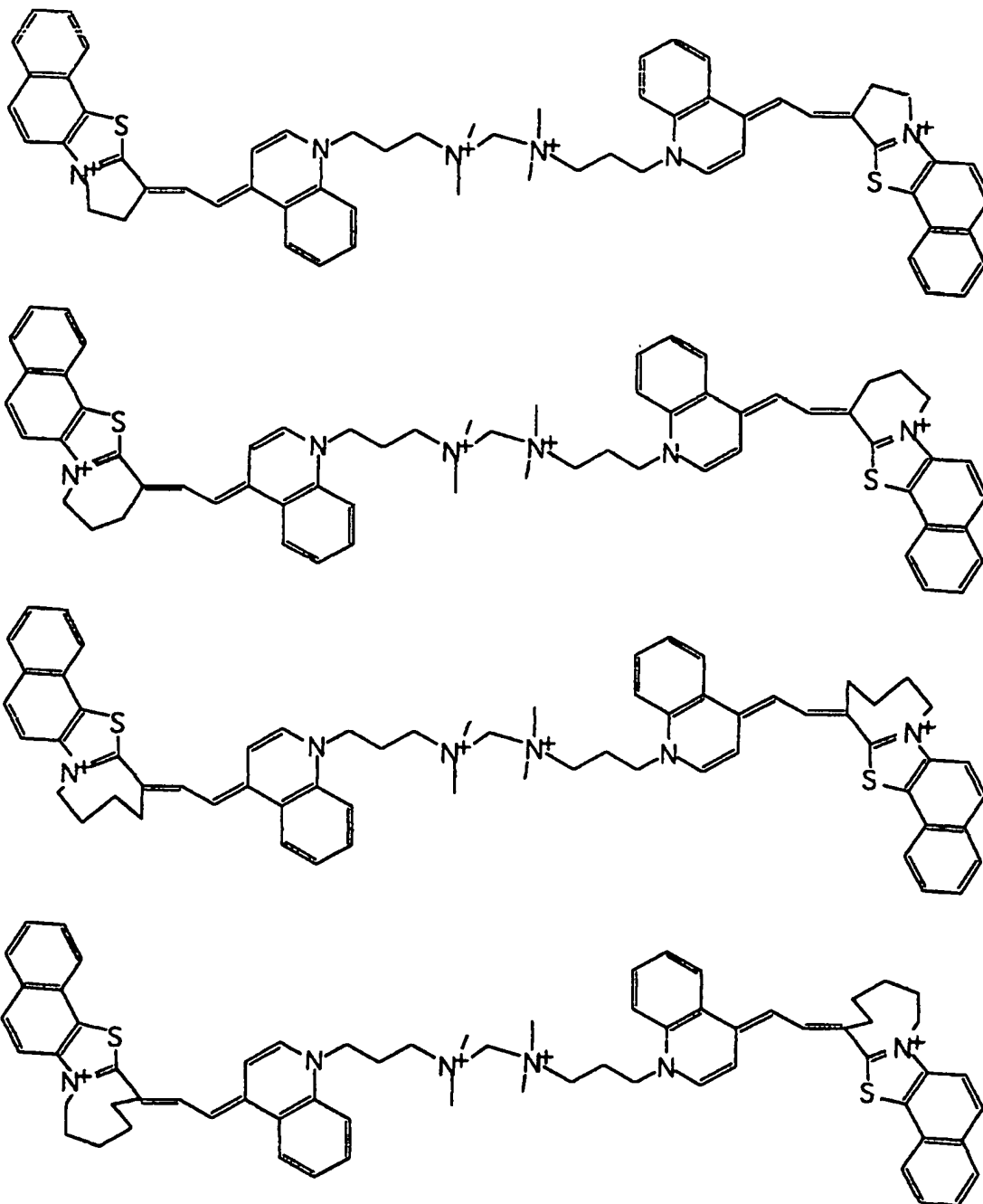


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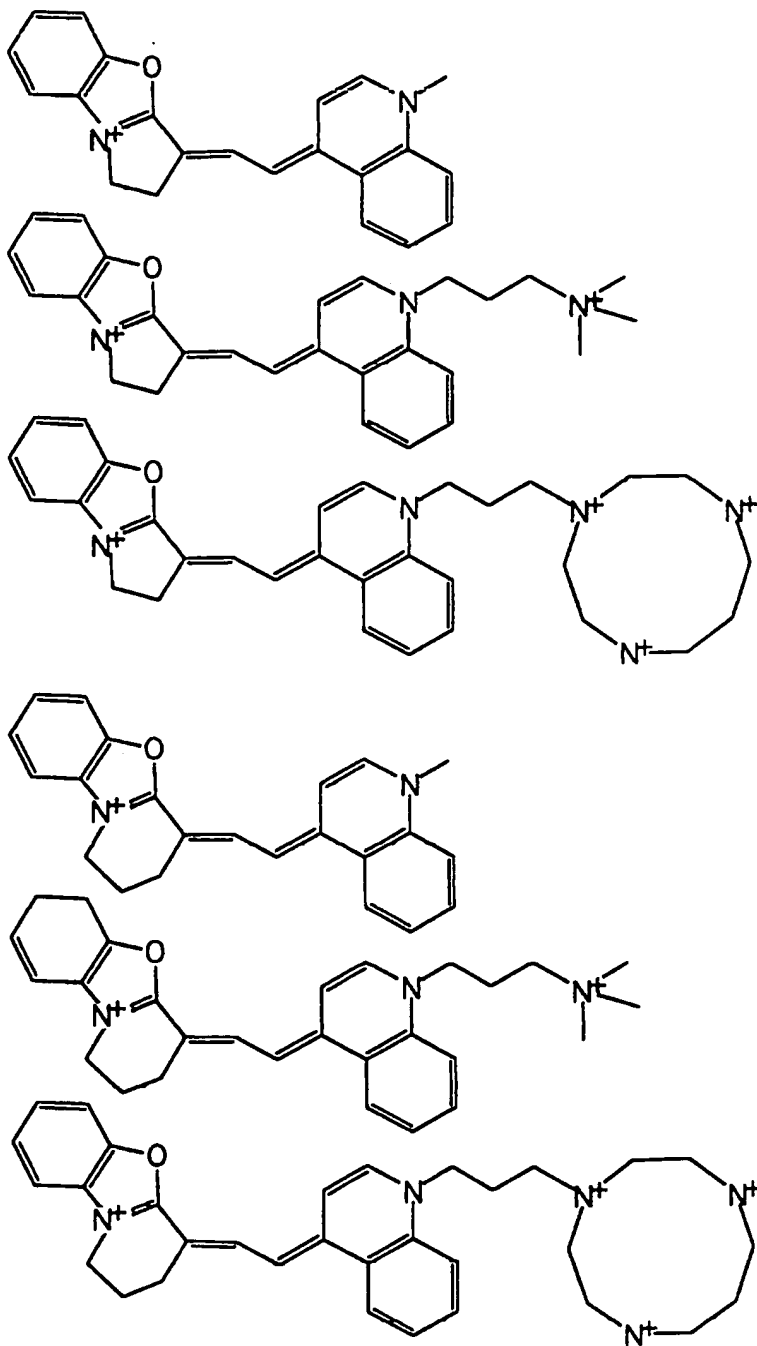


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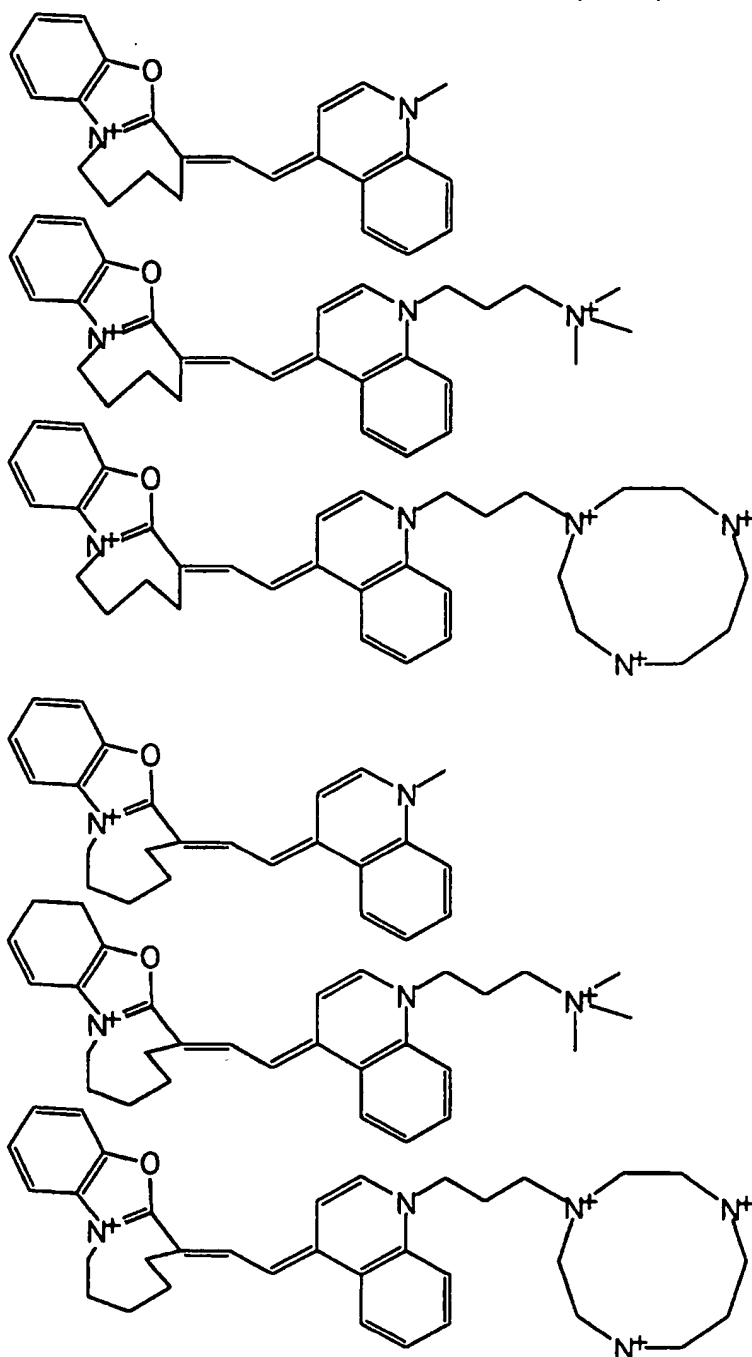
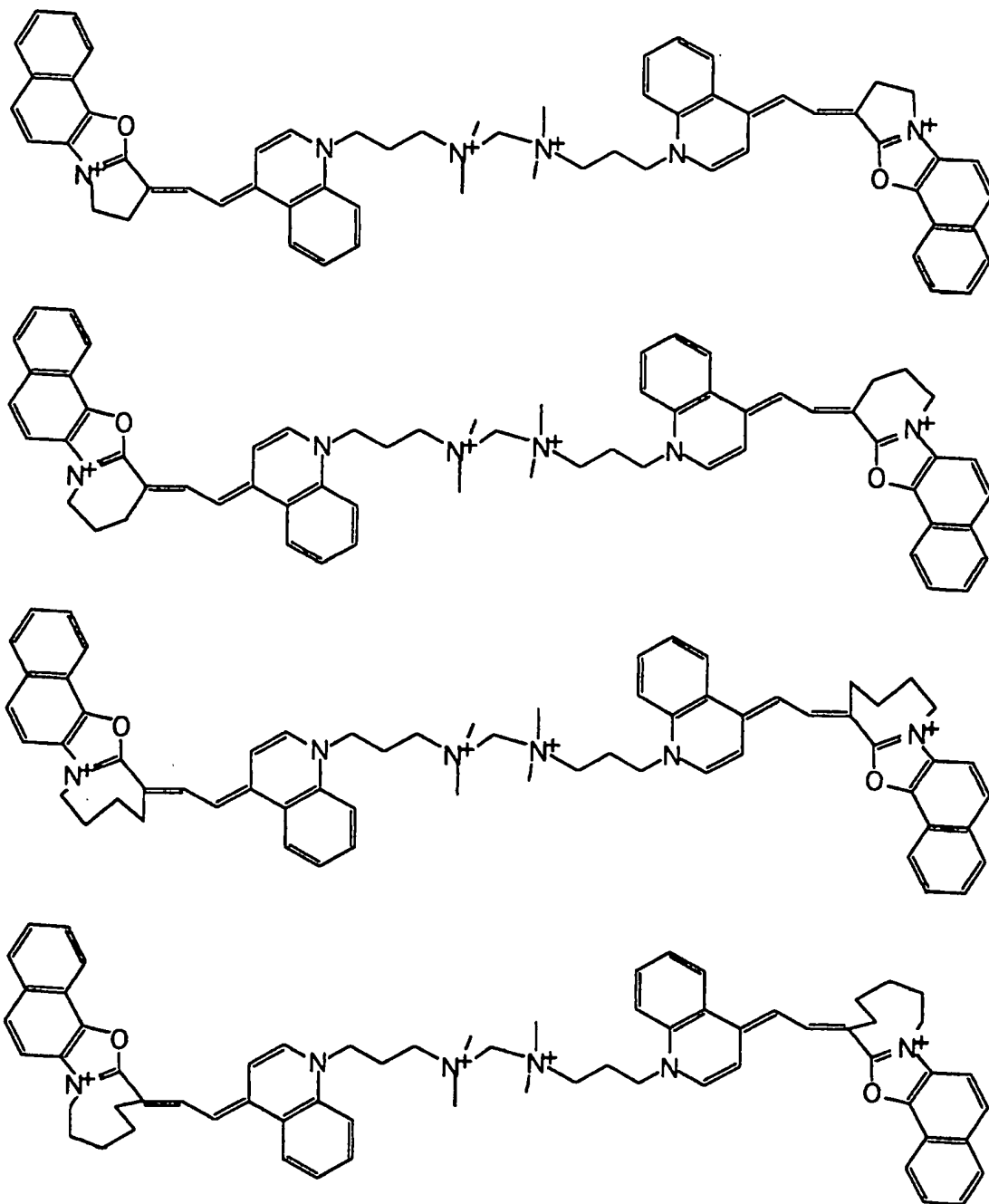
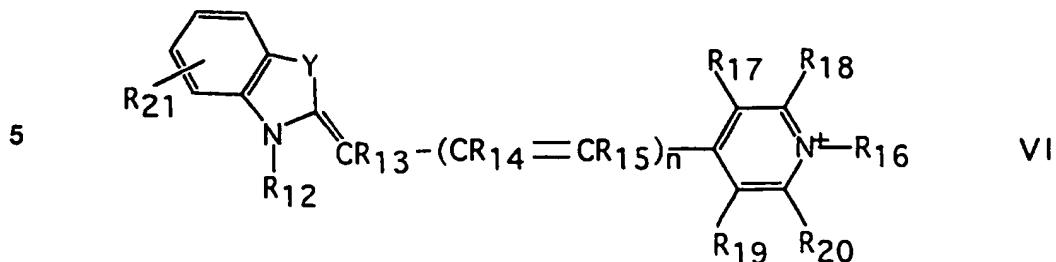


TABLE 2 (cont.)



The present invention also relates to fluorescent cyanine dyes having a positively charged substituent attached to the positively charged nitrogen on the benzothiazole portion of the cyanine dye. These fluorescent cyanine dyes are represented by General Formula VI



where

n is 0, 1 or 2;

Y may be either S or O;

R₁₂ is a positively charged alkyl substituent, more preferably a positively charged aminoalkyl substituent;

R₁₃, R₁₄ and R₁₅ may each independently be either hydrogen, C₁ - C₁₀ alkyl, C₁ - C₁₀ alkoxy, or C₁ - C₁₀ alkylthio;

R₁₂ and R₁₃ may optionally be taken together to form a 5, 6, 7 or 8 membered ring;

R₁₆ may be a C₁ - C₃₀ alkyl, preferably substituted with one or more polar substituents which preferably includes one or more positively charged atoms, or a cyclized fluorescent cyanine dye of the present invention, i.e., where R₁₆ is a linker between two cyclized fluorescent cyanine dyes;

R₁₇ and R₁₈ may each independently be either H or C₁₋₁₀ alkyl, or may be taken together to form a 5 or 6 membered ring, most

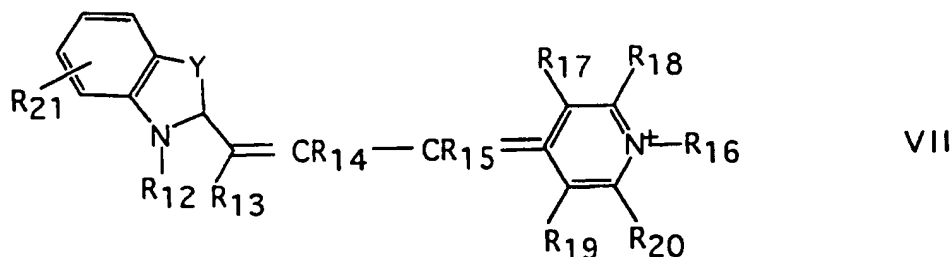
preferably a 5 or 6 membered aromatic ring, optionally substituted with C₁₋₆ alkyl or C₁ - C₁₀ alkoxy groups;

R₁₉ and R₂₀ may each independently be either H or C₁₋₁₀ alkyl, or may be taken together to form a 5 or 6 membered ring, most preferably a 5 or 6 membered aromatic ring, optionally substituted with C₁₋₆ alkyl or C₁ - C₁₀ alkoxy groups; and

R₂₁ may be either H, C₁₋₆ alkyl, C₁ - C₁₀ alkoxy or a fused benzene.

As used above, alkyl and alkoxy refer to any substituent having a carbon backbone having the specified range of carbon atoms. The carbon backbone may form a straight chain, may be branched or may be cyclic. The alkyl and alkoxy groups may be optionally substituted by a wide variety of substituents including, for example, alcohols, amines, thiols, phosphates, halides, ethers, esters, ketones, aldehydes, carboxylic acids, amides, cycloalkyls, and aromatic rings.

With regard to n, it is noted that n may equal 1. Accordingly, an embodiment of the present invention includes cyanine dyes having the General Formula VII (i.e. where n = 1)



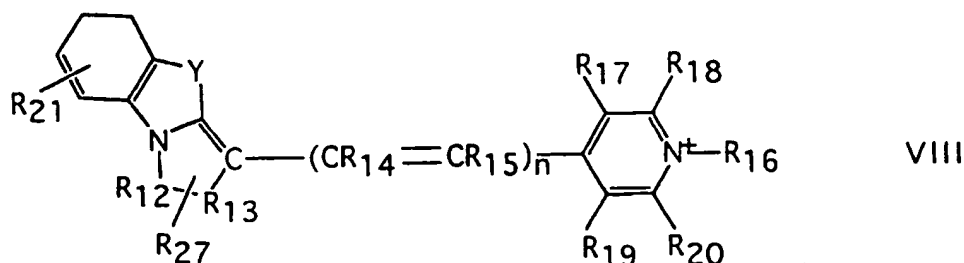
where Y, R₁₂, R₁₃, R₁₄, R₁₅, R₁₆, R₁₇, R₁₈, R₁₉, R₂₀ and R₂₁ are as specified above.

With regard to dyes having General Formula VI or VII, Y may be either S or O and is most preferably S.

R_{12} can be an aminoalkyl chain containing a backbone of 3-42 carbons and 1-5 positively charged nitrogen atoms as described in U.S. Patent No. 5,321,130 to Yue, et al. which is incorporated herein by reference. In addition to the positively charged substituents described in U.S. Patent No. 5,321,130, R_{12} is also intended to include aminoalkyl chains including a positively charged cyclic aminoalkyl group having 1-5 positively charged nitrogen atoms.

In a preferred embodiment, R_{12} has the general formula $-R_{28}N(R_{29}R_{30}R_{31})$ where R_{28} is a C_{1-3} alkyl and R_{29} , R_{30} , and R_{31} are each independently a C_{1-10} alkyl.

In an alternate preferred embodiment, R_{12} and R_{13} are taken together to form a 5, 6, 7 or 8 membered ring where the ring includes a positively charged alkyl substituent, more preferably an aminoalkyl chain containing a backbone of 3-42 carbons and 1-5 positively charged nitrogen atoms as described in U.S. Patent No. 5,321,130 to Yue, et al. Dyes of this embodiment may be generally represented by General Formula VIII



where R_{12} and R_{13} represents the atoms necessary to form a 5, 6, 7 or 8 membered ring and R_{27} is a positively charged substituent, as specified above with regard to R_{12} , which may be attached to any atom used to form the 5, 6, 7 or 8 membered ring as represented by R_{12} and R_{13} . In this regard, these dyes are equivalent to the dyes described above having the General Formula V.

R_{14} and R_{15} may each independently be either hydrogen, C_1 - C_{10} alkyl, C_1 - C_{10} alkylthio, and are preferably H.

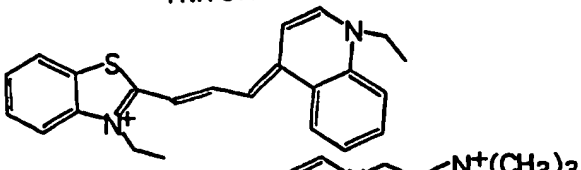
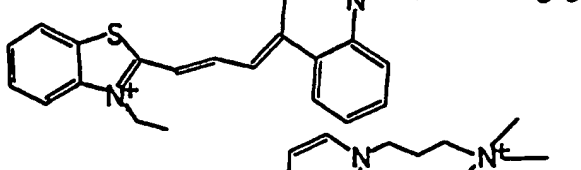
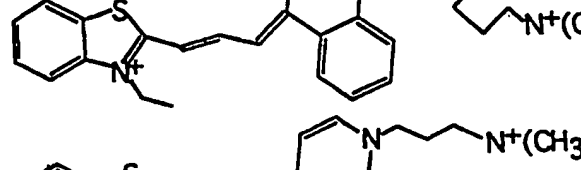
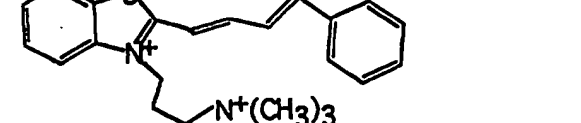
R_{16} may be a C_1 - C_{30} alkyl. Since DNA and RNA to which the cyclized cyanine dyes bind contain negatively charged phosphate backbones, it is preferred that R_{16} be substituted with one or more polar substituents. It is most preferred that R_{16} include one or more positively charged atoms in the polar substituent, such as is specified with regard to R_{12} above.

The cyanine dyes according to General Formula VI, i.e., dyes where a positively charged substituent is positioned off the nitrogen of the benzothiazole portion of the dye, provide the advantage over previous cyanine dyes of exhibiting a significantly larger net fluorescence enhancement with DNA than cyanine dyes where a positively charged substituent is positioned at R_{16} alone.

The use of intercalating dyes for staining cell nuclei requires that the dye itself be membrane-permeable or that a membrane permeabilizing step be incorporated into the sample preparation. Methods for enabling charged molecules and very large molecules into cells include the use of chemicals, such as digitonin, freeze-thaw cell lysis steps, or the use of non-ionic detergents such as TRITON X-100. For speed and simplicity, it is preferred to add approximately 9mM TRITON X-100.

The presence of a detergent solution (TRITON X-100) causes significant fluorescence enhancement of the dyes as compared to in PBS buffer. An increase in detergent-enhanced fluorescence ($F_{\text{TRITON}}/F_{\text{PBS}}$) has the effect of decreasing the net DNA enhanced fluorescence over detergent-enhanced background fluorescence ($F_{\text{DNA}}/F_{\text{TRITON}}$). The detergent-enhanced fluorescence is believed to increase with increasing hydrophobicity.

TABLE 3: Fluorescence Ratios of Dyes in Buffer,
TRITON X-100 and DNA Solutions

		ETriton/ EPBS	EDNA/ EPBS	EDNA/ ETRITON	
5		6 *	94	200	3
		5 *	12	100	8
		7 *	10	70	7
		8	1.8	70	40

* Compounds 5 and 7 are taught in U.S. Patent No. 5,321,130 to Yue, et al.

* Compound 6 is taught in U.S. Patent No. 4,957,870 to Lee, et al.

Fluorescence enhancement of the dyes upon binding to an excess of DNA was found to be fairly constant regardless of how the quinolinium ring side chain was modified (R_{1a}). Advantageously, however, it was found that inclusion of a positively charged substituent off the positively charged nitrogen of the benzothiazole portion of the dye (General Formula VI) causes the dye to exhibit a significantly larger net DNA-enhancement than the positioning of a positively charged substituent at R_{1a} alone. As a result, smaller concentrations of nucleic acids can be detected using cyanine dyes having General Formula VI.

For example, Table 3 compares the fluorescence ratios of dyes in a saline buffer, a detergent (TRITON X-100) and in a DNA solution. Dye solutions (1.0 μ M) were prepared in phosphate buffered saline (PBS), in PBS with TRITON X-100 (9mM), and in PBS with double-stranded DNA (100 μ M).

5 Table 3 shows the effect of various side chains on the fluorescence background in TRITON X-100 (9mM). As illustrated in Table 3, the net DNA enhanced fluorescence over detergent-enhanced background fluorescence (F_{DNA}/F_{TRITON}) was found to be a factor of 5 greater in dye 8 than in dye 7. This result is unexpected since the net charge of 3+ is the same for both dyes 7 and
10 8. It appears that both the location and quantity of charges affect the fluorescence enhancement of the dyes.

The cyanine dyes according to General Formula VI preferably absorb light at a wavelength of at least about 640 nm, more preferably at least about 649 nm and emit fluorescence at a wavelength of at least about 650 nm, more
15 preferably at least about 663 nm. The cyanine dyes also preferably have a positive Stoke's shift ($\lambda_{Emission} - \lambda_{Abs.}$) of at least 11 nm.

Table 4 provides examples of some of the preferred cyanine dyes having General Formula VI. It should be understood, however, that the dyes listed in Table 4 are intended only to exemplify the cyanine dyes of the present
20 invention and are not intended to be limiting.

TABLE 4

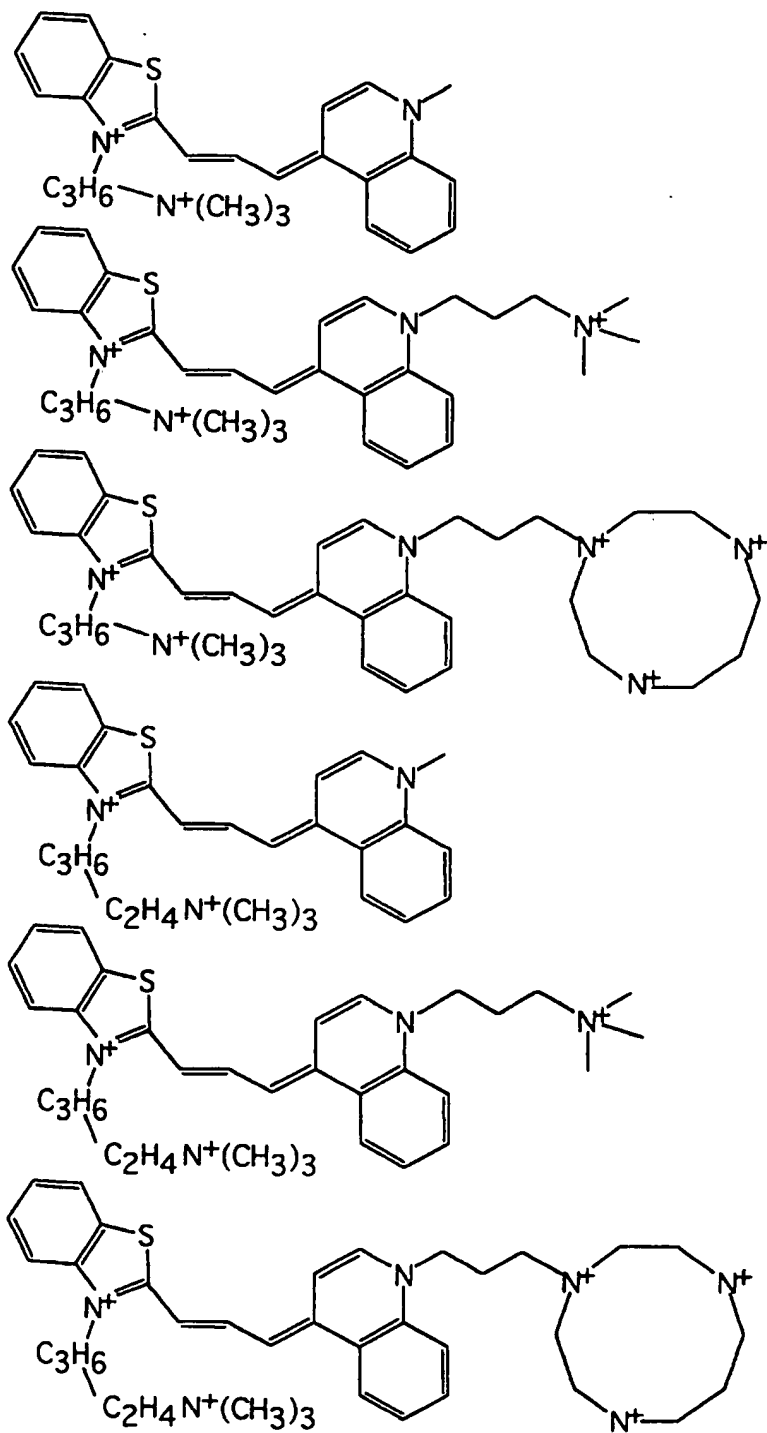


TABLE 4 (cont.)

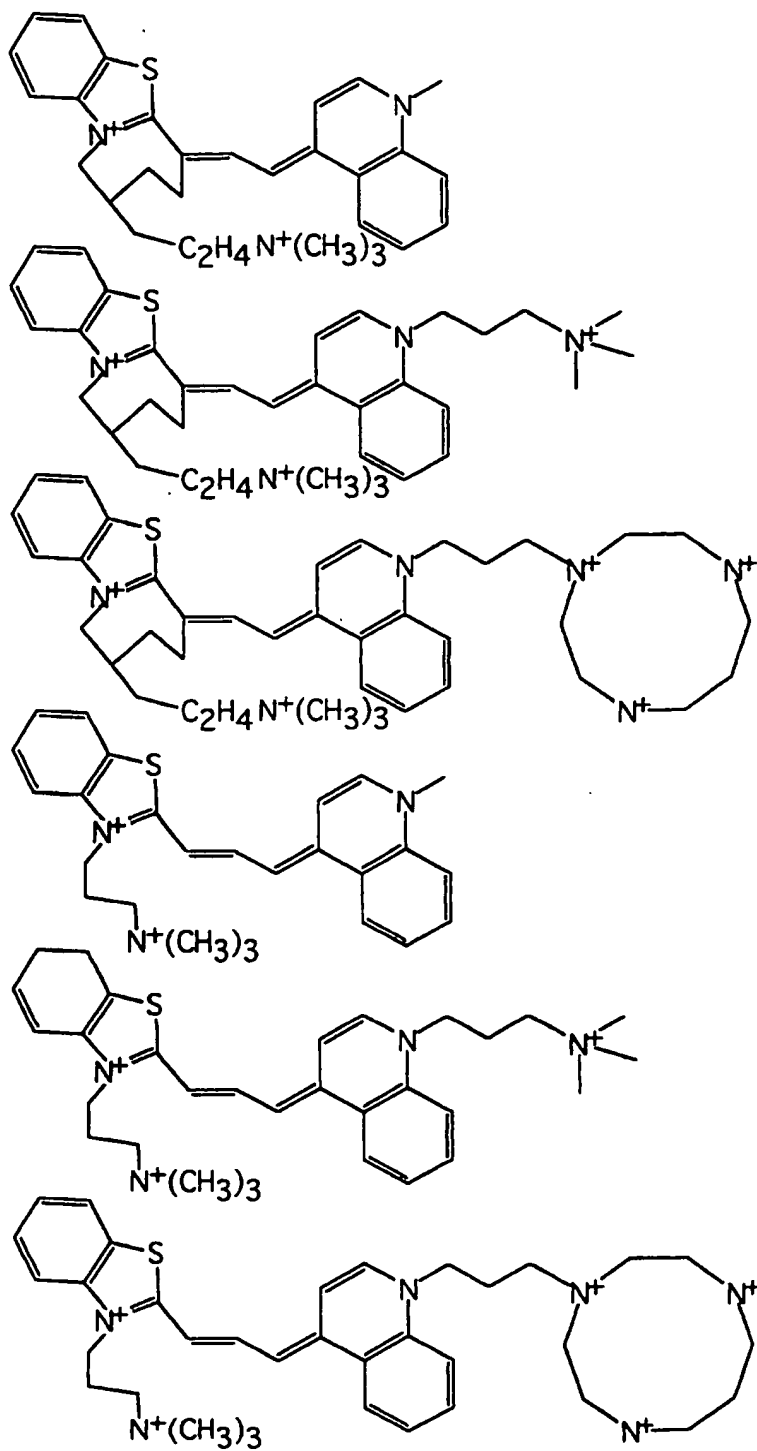


TABLE 4 (cont.)

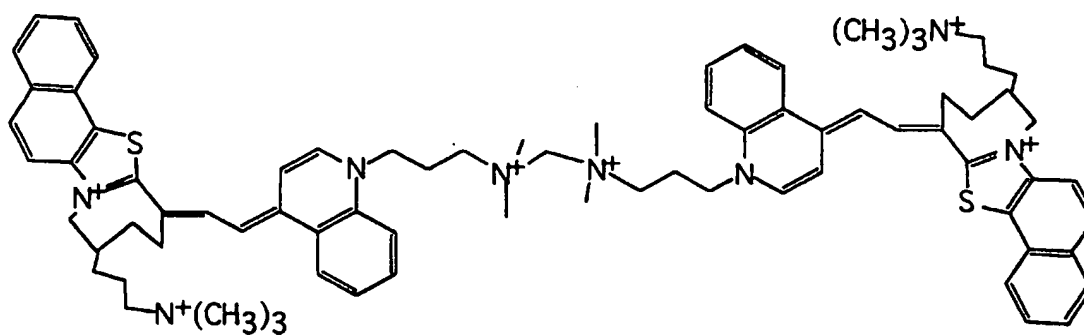
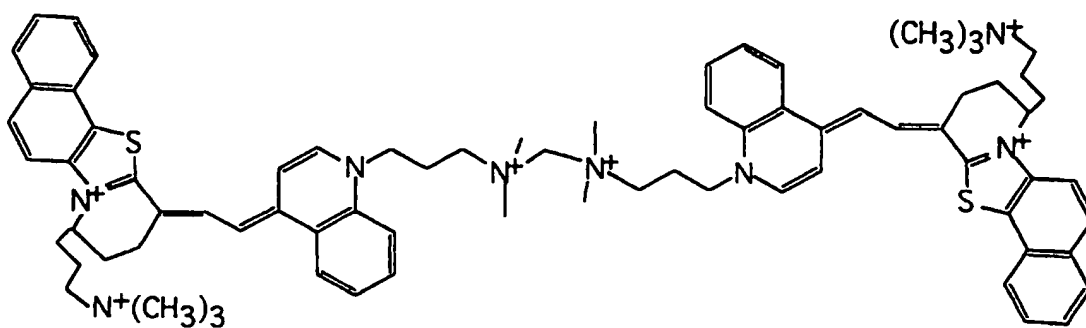
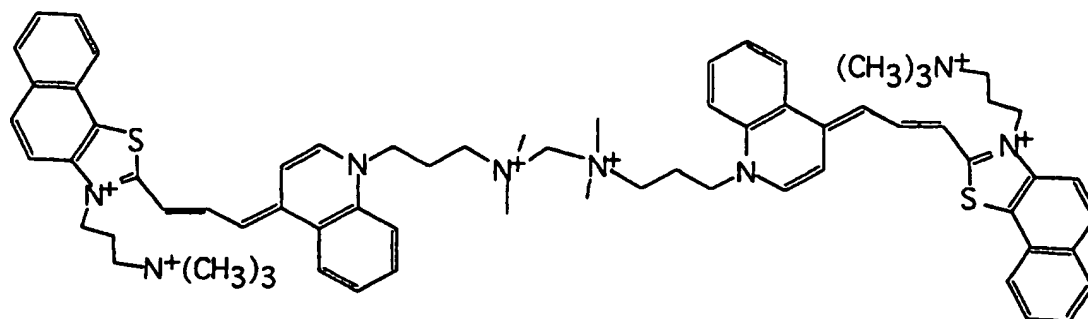


TABLE 4 (cont.)

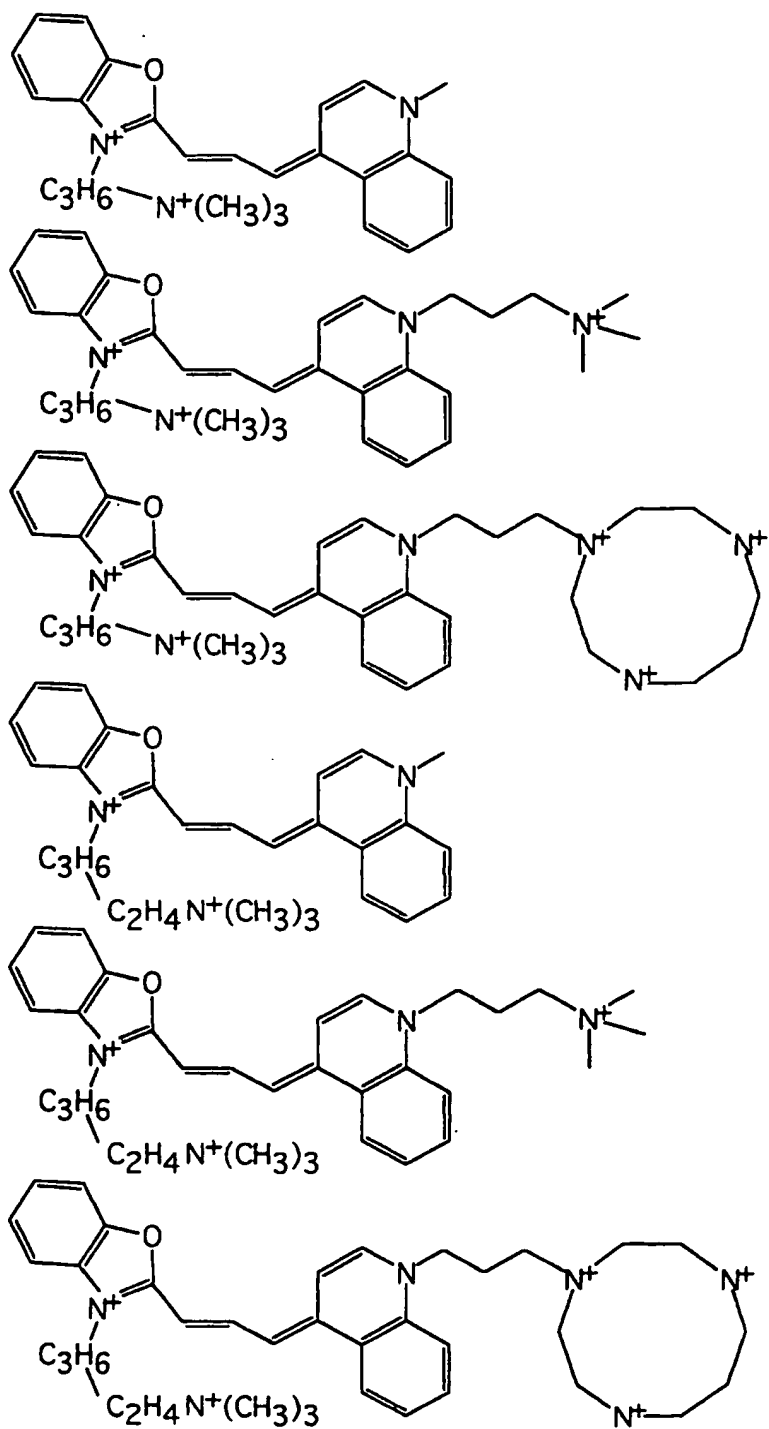


TABLE 4 (cont.)

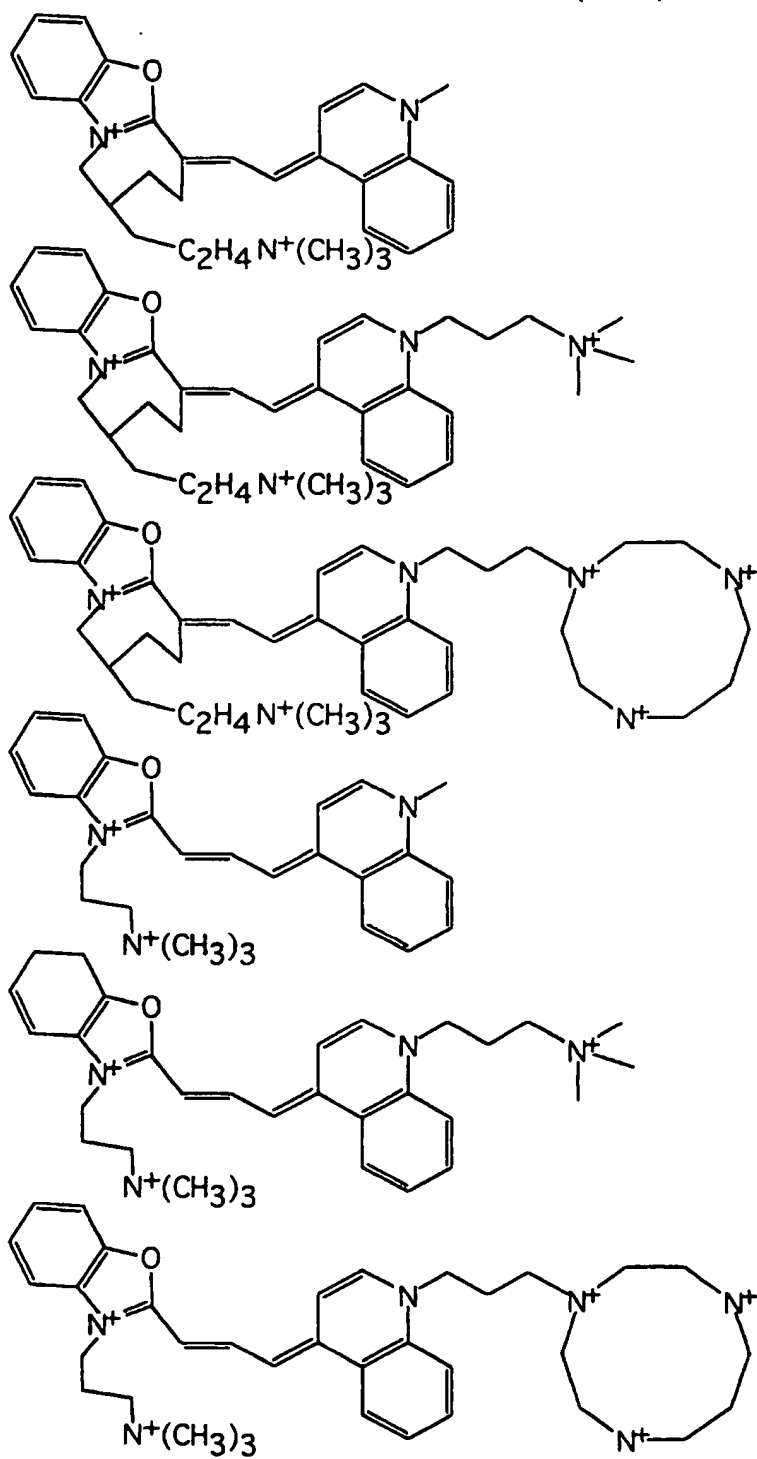
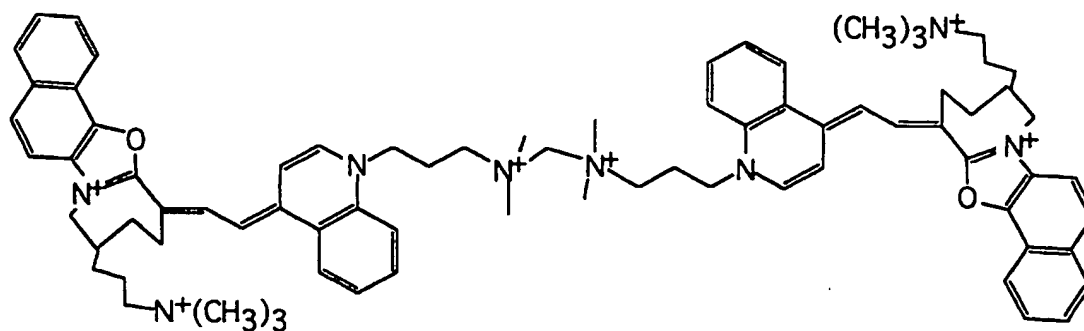
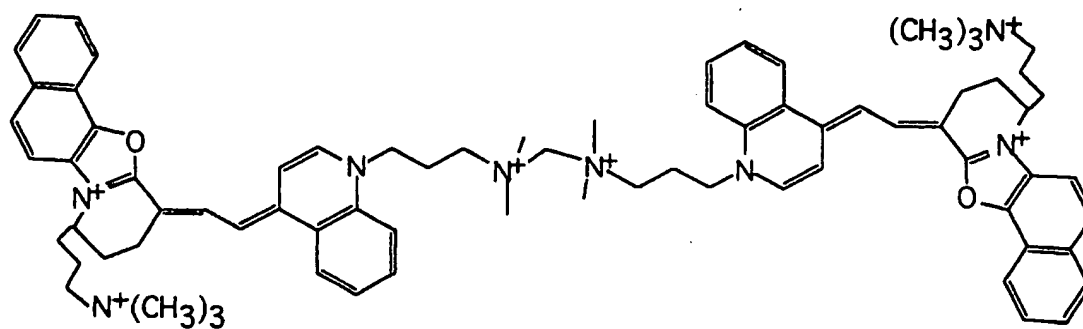
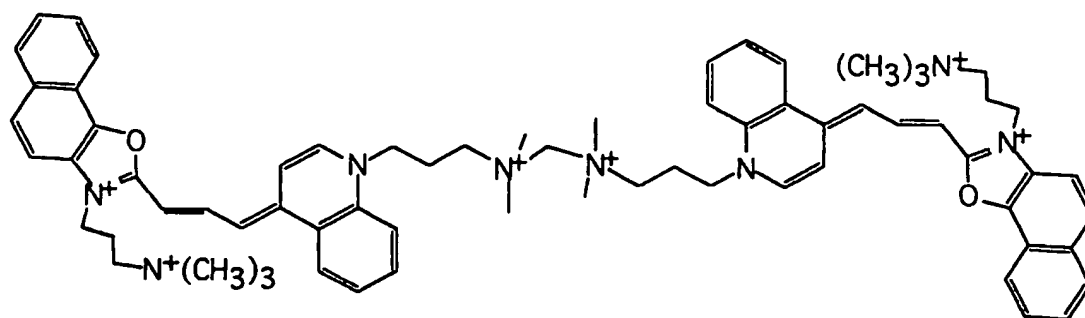


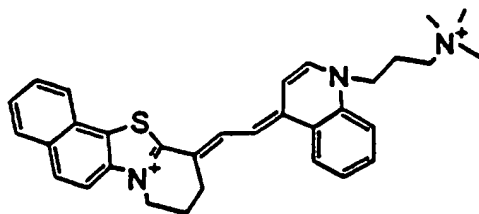
TABLE 4 (cont.)



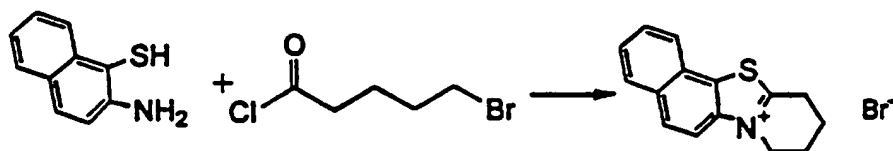
The present invention also relates to the use of the cyanine dyes having General Formulas I, II, IV, V, VI, VII or VIII to form compositions for detecting the presence of nucleic acids in a sample. In general, the compositions include a cyanine dye according to the present invention non-covalently bound to a nucleic acid, i.e., DNA or RNA.

The fluorescence of the cyanine dyes of the present invention significantly increase when bound to a nucleic acid. As a result, it is possible to qualitatively or quantitatively determine the presence of nucleic acids in a sample by monitoring the change in the fluorescence intensity of the dye at a wavelength corresponding to the composition of the dye bound to the nucleic acids. Use of cyanine dyes in general for detecting the presence of nucleic acids in a sample is known in the art. A discussion regarding the use of cyanine dyes to detect the presence of nucleic acids in a sample is provided in U.S. Patent No. 5,321,130 to Yue, et al. which is incorporated herein by reference.

The present invention also relates to a method for detecting nucleic acids by contacting the nucleic acids with a cyanine dye of the present invention. According to the method, a sample of nucleic acids are contacted with a cyanine dye of the present invention in order to form the composition of a cyanine dye non-covalently bound to a nucleic acid sequence. After the dye-nucleic acid sequence composition is formed, the bound dye is exposed to light having a wavelength near an absorbance maximum of the dye when bound to a nucleic acid sequence. The resulting fluorescence emission of the dye is then detected in order to qualitatively or quantitatively determine the presence of nucleic acids in the sample.

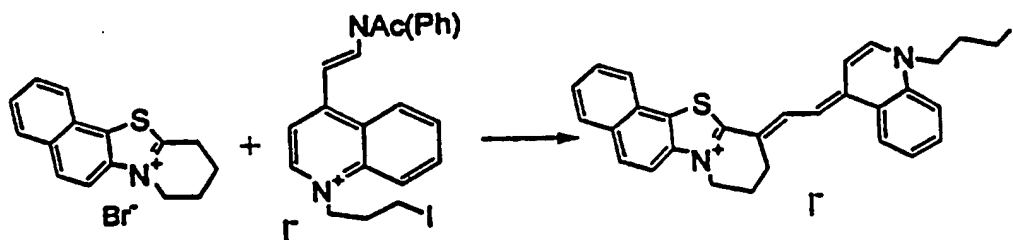
Example 1: Preparation of Compound 4

1a. Preparation of 2,3-Tetramethylenenaphth[2,1-d]thiazolium
Bromide



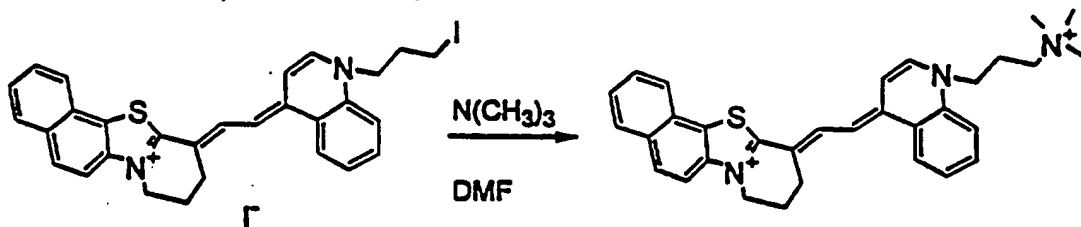
2-Aminonaphthalene-1-thiol was prepared by the method of Ambroggi, et al. (Ambroggi, V.; Grandolini, G.; Peroli, L.; Rossi, C. *Synthesis*, 1992, 7, 656-8.) 2-Aminonaphthalene-1-thiol (0.14 g, 0.8 mmol) and bromovaleryl chloride (0.48 g, 2.4 mmol) were combined and heated to 100° for 1 h, then to 50°C overnight. The resulting solid was washed with acetone and air-dried to provide a white solid (0.16 g, 0.5 mmol, 60% yield).

1b. Preparation of IodoNAP6



4-(2'-Acetanilidovinyl)-1'-(3'-Iodopropyl)-quinolinium iodide (prepared by the general method of Brooker, et al. *J. Am. Chem. Soc.* 1941, 63, 3192-3203; 32 mg, 63 μ mol), 2,3-tetramethylenenaphth[2,1-d]thiazolium bromide (20 mg, 63 μ mol), triethylamine (40 μ L) and ethanol (1 mL) were combined and refluxed for 20 min. The dark blue solid was recrystallized sequentially from isopropanol and ethanol to provide a purple solid (12 mg, 30% yield). HPLC analysis on a C8 reverse-phase column using gradient elution of 40% to 80% acetonitrile vs. 0.1 M triethylammonium acetate buffer showed one major peak at 16 min.

1c. Preparation of Compound 4



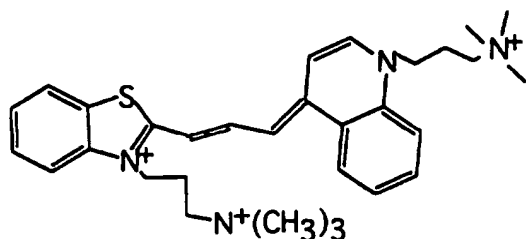
IodoNAP6 (2 mg, 3 μ mol) was dissolved in dimethylformamide. Trimethylamine was bubbled through the solution. The reaction was monitored by thin layer chromatography on silica gel with methanol as the eluant. The R_f values of IodoNAP6 and compound 4 were 0.5 and zero, respectively. After 30 min, reaction was complete. The solvent was evaporated and the residue partitioned between methylene chloride (CH_2Cl_2) and water. The aqueous layer was washed with 2 x 1 mL CH_2Cl_2 and concentrated to dryness. HPLC

analysis with the same gradient that was used with iodoNAP6 showed one broad peak at 7.2 min with no apparent starting material. The absorbance maximum of compound 4 in methanol was at 667 nm.

5

Example 2: Preparation of Compound 8

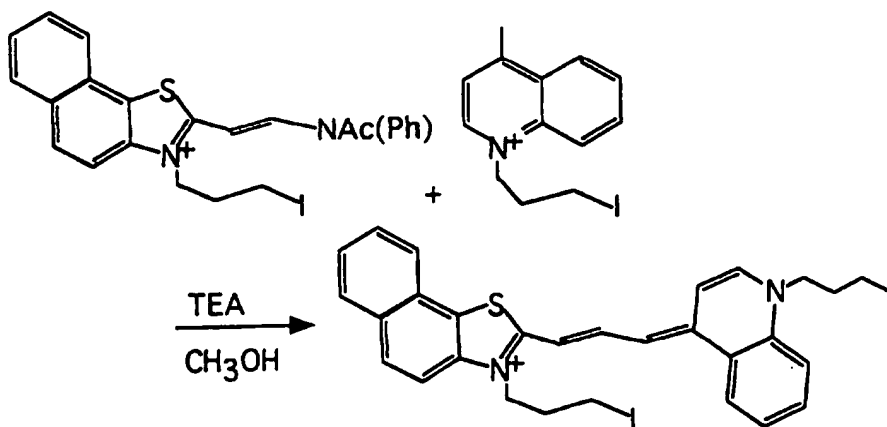
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15

2a. Preparation of 1',1''-(3',3''Bisiodopropyl)-thia-4-carbocyanine iodide

20

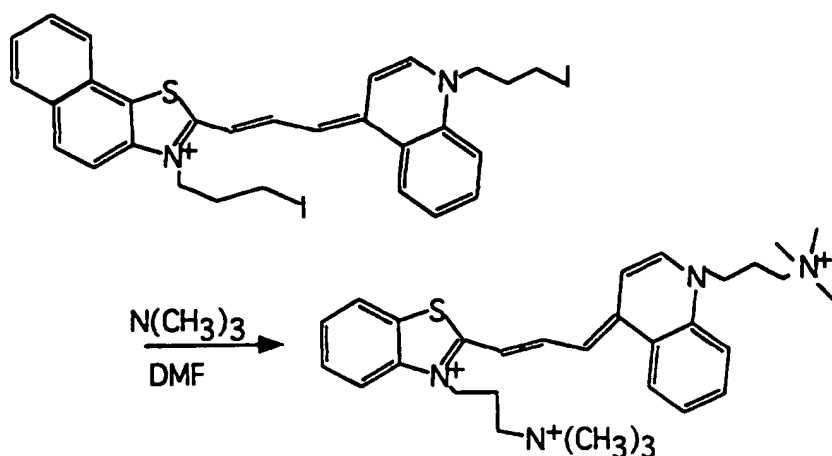


25

1'3(e'-Iodopropyl)-2-(2''-acentanilidovinyl)-benzothiazium iodide (15 mg, 26 μ mol), 1'-3(3'-iodopropyl)-quinolinium iodide (15 mg, 34 μ mol), triethylamine (50 μ L) and methanol (1 mL) were combined at room temperature. A blue precipitate formed immediately. The reaction mixture was centrifuged and the residue washed with methanol and isopropanol and air-dried to provide a dark solid (15 mg, 20 μ mol, 77% yield).

35

2b. Preparation of Compound 8

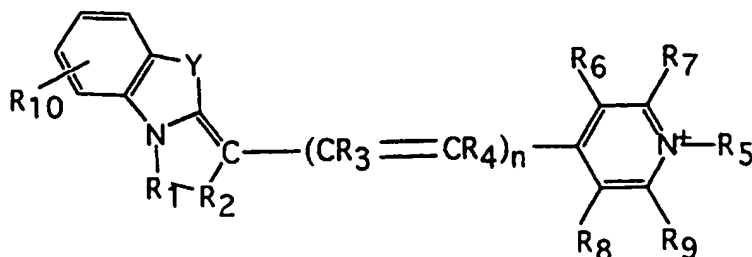


1', 1''-(3', 3'-Bisiodopropyl)-thia-4-carbocyanine iodide (15 mg, 20 μ mol) was dissolved in DMF and trimethylamine bubbled through the solution. The reaction progress was monitored by TLC on reverse-phase plates with 1:1 dimethylformamide:4 M NaCl as eluant. The R_f 's of the bisiodo starting material and the bisammonium salt were 0 and 0.8, respectively. The intermediate monoammonium salts could also be resolved, at R_f 's of 0.7 and 0.6. After 30 min the reaction was complete. The solvent was evaporated. The absorbance maximum of compound 8 in DMSO was at 639 nm.

While the present invention is disclosed by reference to the preferred embodiments and examples detailed above, it is to be understood that these examples are intended in an illustrative rather than limiting sense, as it is contemplated that modifications will readily occur to those skilled in the art, which modifications will be within the spirit of the invention and the scope of the appended claims.

What is claimed is:

1. A composition comprising: a cyclized fluorescent cyanine dye noncovalently bound to a nucleic acid polymer, the cyclized cyanine dye having the general formula



where:

n is 0, 1 or 2;

Y is selected from the group consisting of S or O;

R_1 and R_2 are taken together to form a 5, 6, 7 or 8 membered ring;

R_3 and R_4 are each independently selected from the group consisting of hydrogen, $C_1 - C_{10}$ alkyl, $C_1 - C_{10}$ alkoxy and $C_1 - C_{10}$ alkylthio;

R_5 is a $C_1 - C_{50}$ alkyl;

R_6 and R_7 are each independently selected from the group consisting of H, and C_{1-10} alkyl, or where R_6 and R_7 together to form a 5 or 6 membered ring;

R_8 and R_9 are each independently selected from the group consisting of H and C_{1-10} alkyl, or where R_8 and R_9 together to form a 5 or 6 membered ring; and

R_{10} is selected from the group consisting of H, C_{1-6} alkyl, and a fused benzene.

- 1 2. The composition according to claim 1 wherein dye includes a
2 positively charged substituent attached to one of the atoms represented by R₁
3 and R₂ forming the 5, 6, 7 or 8 membered ring.
- 1 3. The composition according to claim 2 wherein the positively
2 charged substituent includes an aminoalkyl group.
- 1 4. The composition according to claim 3 wherein the aminoalkyl
2 group includes a positively charged cyclic aminoalkyl group having a 1-5
3 positively charged nitrogen atoms.
- 1 5. The composition according to claim 3 wherein the aminoalkyl
2 group has the general formula -R₂₈N(R₂₉R₃₀R₃₁) where R₂₈ is a C₁₋₃ alkyl and
3 R₂₉, R₃₀ and R₃₁ are each independently a C₁₋₁₀ alkyl.
- 1 6. The composition of claim 1 wherein R₁ and R₂ are taken together
2 to form a 7 or 8 membered ring.
- 1 7. The composition of claim 1 wherein R₃ is substituted by at least
2 one polar substituent.
- 1 8. The composition of claim 7 wherein R₃ is substituted by at least
2 one positively charged atom.
- 1 9. The composition of claim 8 wherein R₃ is an aminoalkyl chain
2 containing a backbone of two to about 42 carbons and 1-5 positively charged
3 nitrogens intermittently or equally spaced within the backbone, such that there
4 are at least two carbons between sequential nitrogens.
- 1 10. The composition of claim 9 wherein R₃ forms a 6-33 membered
2 ring.

1 11. The composition of claim 10 wherein R_6 is a positively charged
2 cyclic aminoalkyl group having 1-5 positively charged nitrogen atoms.

1 12. The composition of claim 1 wherein $n = 1$.

1 13. The composition of claim 1 wherein Y is S.

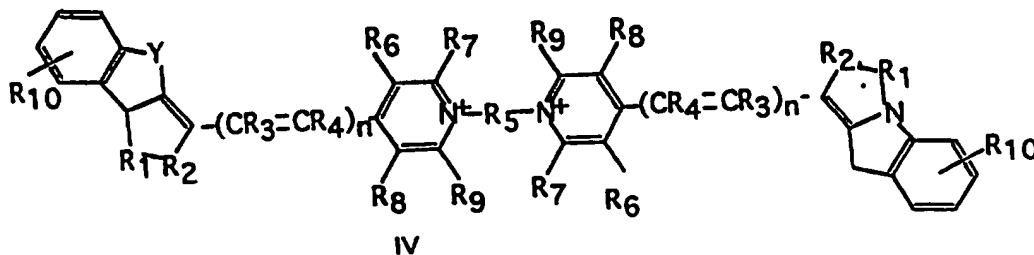
1 14. The composition of claim 1 wherein R_6 and R_7 or R_6 and R_8 are
2 each H.

1 15. The composition of claim 14 wherein R_6 and R_7 or R_6 and R_8 are
2 taken together to form a 5 or 6 membered ring.

1 16. The composition of claim 15 wherein the ring formed by R_6 and
2 R_7 or R_6 and R_8 is a 6 membered aromatic ring.

1 17. The composition of claim 1 wherein R_{10} is a fused benzene.

1 18. A composition comprising: a cyclized fluorescent cyanine dye
2 noncovalently bound to a nucleic acid polymer, the cyclized cyanine dye
3 having the general formula



3 where:

4 n is 0, 1 or 2;

5 Y is selected from the group consisting of S or O;
6 R₁ and R₂ are taken together to form a 5, 6, 7 or 8 membered ring;
7 R₃ and R₄ are each independently selected from the group consisting of
8 hydrogen, C₁ - C₁₀ alkyl, C₁ - C₁₀ alkoxy and C₁ - C₁₀ alkylthio;
9 R₅ is a C₁ - C₃₀ alkyl;
10 R₆ and R₇ are each independently selected from the group consisting of
11 H, and C₁₋₁₀ alkyl, or where R₆ and R₇ together to form a 5 or 6 membered ring;
12 and
13 R₁₀ is selected from the group consisting of H, C₁₋₆ alkyl, C₁ - C₁₀ alkoxy
14 and a fused benzene.

1 19. The composition according to claim 18 wherein dye includes a
2 positively charged substituent attached to one of the atoms represented by R₁
3 and R₂ forming th 5, 6, 7 or 8 membered ring.

1 20. The composition of claim 18 wherein R₁ and R₂ are taken
2 together to form a 7 or 8 membered ring.

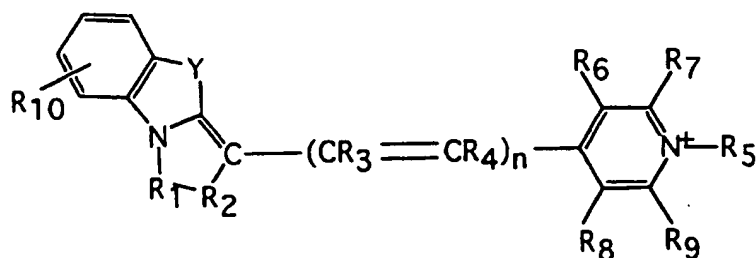
1 21. The composition of claim 18 wherein R₅ is substituted by at least
2 one polar substituent.

1 22. The composition of claim 21 wherein R₅ is substituted by at least
2 one positively charged atom.

1 23. The composition of claim 22 wherein R₅ is an aminoalkyl chain
2 containing a backbone of two to about 42 carbons and 1-5 positively charged
3 nitrogens intermittently or equally spaced within the backbone, such that there
4 are at least two carbons between sequential nitrogens.

1 24. The composition of claim 18 wherein
2 n = 1.

- 1 25. The composition of claim 18 wherein Y is S.
- 1 26. The composition of claim 18 wherein R₆ and R₇ or R₈ and R₉ are
2 each H.
- 1 27. The composition of claim 26 wherein R₆ and R₇ or R₈ and R₉ are
2 taken together to form a 5 or 6 membered ring.
- 1 28. The composition of claim 27 wherein the ring formed by R₆ and
2 R₇ or R₈ and R₉ is a 6 membered aromatic ring.
- 1 29. The composition of claim 18 wherein R₁₀ is a fused benzene.
- 1 30. A method for detecting a nucleic acid polymer comprising:
2 contacting a nucleic acid sequence with a cyclized fluorescent cyanine
3 dye to form a noncovalently bound dye-nucleic acid composition, the cyclized
4 fluorescent cyanine dye having the general formula



- 5 where
- 6 n is 0, 1 or 2;
- 7 Y is selected from the group consisting of S or O;

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8 R₁ and R₂ are taken together to form a 5, 6, 7 or 8 membered
9 ring;

10 R₃ and R₄ are each independently selected from the group
11 consisting of hydrogen, C₁ - C₁₀ alkyl, C₁ - C₁₀ alkoxy and C₁ - C₁₀ alkylthio;

12 R₅ is a C₁ - C₈₀ alkyl;

13 R₆ and R₇ are each independently selected from the group
14 consisting of H and C₁₋₁₀ alkyl, or where R₆ and R₇ together to form a 5 or 6
15 membered ring;

16 R₈ and R₉ are each independently selected from the group
17 consisting of H and C₁₋₁₀ alkyl, or where R₈ and R₉ together to form a 5 or 6
18 membered ring; and

19 R₁₀ is selected from the group consisting of H, C₁₋₄ alkyl, C₁ - C₁₀
20 alkoxy and a fused benzene;

21 exposing the cyclized fluorescent cyanine dye bound to the nucleic acid
22 polymer to light, the cyclized fluorescent cyanine dye absorbing the light and
23 producing a fluorescence emission; and

24 detecting the fluorescence emission.

1 31. The method according to claim 30 wherein dye includes a
2 positively charged substituent attached to one of the atoms represented by R₁
3 and R₂ forming the 5, 6, 7 or 8 membered ring.

1 32. The method according to claim 31 wherein dye includes a
2 positively charged substituent attached to one of the atoms represented by R₁
3 and R₂ forming the 5, 6, 7 or 8 membered ring.

1 33. The method according to claim 32 wherein dye includes a
2 positively charged substituent attached to one of the atoms represented by R₁
3 and R₂ forming the 5, 6, 7 or 8 membered ring.

1 34. The method according to claim 33 wherein dye includes a
2 positively charged substituent attached to one of the atoms represented by R₁
3 and R₂ forming the 5, 6, 7 or 8 membered ring.

1 35. The method of claim 30 wherein R₁ and R₂ are taken together to
2 form a 7 or 8 membered ring.

1 36. The method of claim 30 wherein n = 1.

1 37. The method of claim 30 wherein Y is S.

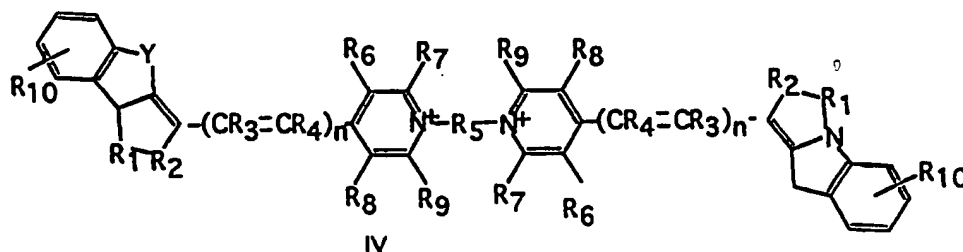
1 38. The method of claim 30 wherein R₆ and R₇ or R₈ and R₉ are each
2 H.

1 39. The method of claim 38 wherein R₆ and R₇ or R₈ and R₉ are taken
2 together to form a 5 or 6 membered ring.

1 40. The method of claim 38 wherein the ring formed by R₆ and R₇ or
2 R₈ and R₉ is a 6 membered aromatic ring.

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41. A method for detecting a nucleic acid polymer comprising:
 contacting a nucleic acid polymer with a cyclized fluorescent cyanine dye
 to form a noncovalently bound dye-nucleic acid polymer composition, the
 cyclized fluorescent cyanine dye having the general formula



where

n is 0, 1 or 2,

Y is selected from the group consisting of S or O;

R_1 and R_2 are taken together to form a 5, 6, 7 or 8 membered ring,

R_3 and R_4 are each independently selected from the group consisting of hydrogen, $C_1 - C_{10}$ alkyl, $C_1 - C_{10}$ alkoxy and $C_1 - C_{10}$ alkylthio,

R_5 is a $C_1 - C_{20}$ alkyl,

R_6 and R_7 are each independently selected from the group consisting of H and C_{1-10} alkyl, or where R_6 and R_7 together to form a 5 or 6 membered ring,

R_8 and R_9 are each independently selected from the group consisting of H and C_{1-10} alkyl, or where R_8 and R_9 together to form a 5 or 6 membered ring and

R_{10} is selected from the group consisting of H, C_{1-6} alkyl, $C_1 - C_{10}$ alkoxy and a fused benzene;

exposing the cyclized fluorescent cyanine dye bound to the nucleic acid polymer to light, the cyclized fluorescent cyanine dye absorbing the light and producing a fluorescence emission; and

detecting the fluorescence emission.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US96/17943

A. CLASSIFICATION OF SUBJECT MATTER

IPC(6) : C12Q 1/68

US CL : 435/6

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 435/6,7.1,91.1,91.2,810; 536/22.1,23.1,24.1,24.3,24.31,24.32,24.33; 935/77,78

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

Dialog search covering CAS, Biotech abs., Medline, WPI
search words: cyanine, dye?, hybridization, and hybridiz?

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	Nucleic Acids Research, Volume 23, Number 7, issued 1995, Rye et al., "Interaction of dimeric intercalating dyes with single-stranded DNA", pages 1215-1222, see especially the abstract.	1-41
A	Nucleic Acids Research, Volume 20, Number 11, issued 1992, Rye et al., "Stable fluorescent complexes of double-stranded DNA with bis-intercalating asymmetric cyanine dyes: properties and applications", pages 2803-2812, see especially the abstract.	1-41
A	US, A, 4,875,762 (KATO ET AL.) 24 October 1989, see entire document.	1-41
A	US, A, 4,622,391 (LORENZ ET AL.) 11 November 1986, see entire document.	1-41

☐ Further documents are listed in the continuation of Box C.
 ☐ See patent family annex.

* Special categories of cited documents:	*T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
A document defining the general state of the art which is not considered to be of particular relevance	*X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
E earlier document published on or after the international filing date	*Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
L document which may throw doubt on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	*A* document member of the same patent family
O document referring to an oral disclosure, use, exhibition or other means	
P document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search

22 DECEMBER 1996

Date of mailing of the international search report

28 JAN 1997

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